THE EUROPEAN DIRECTORATE FOR THE QUALITY OF MEDICINES & HEALTHCARE (EDQM)





Use of RS in specific monographsRS for identification and RS for assay

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"The European Pharmacopoeia"
Dr Jochen Pauwels
EDQM Laboratory Department

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IDENTIFICATION

Identification methods that generally require comparison with an RS

- → infrared absorption spectrophotometry (IR)
- → nuclear magnetic resonance spectrometry (NMR), if spectrum cannot be interpreted
- → chromatographic separation techniques (LC, GC, TLC)

RS strategy

- → substance compliant with corresponding Ph.Eur. monograph
- → in some justified cases, another form is used e.g. salt/hydrate (sample preparation needed)
- → sometimes the identification RS is also used as external standard in the LC/GC assay
- → other uses e.g. system suitability and peak identification are normally to be avoided

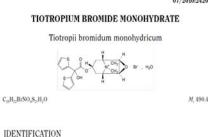
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IDENTIFICATION

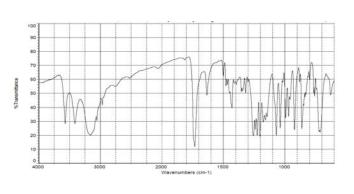
Example IR



A. Infrared absorption spectrophotometry (2.2.24).

Comparison: tiotropium bromide monohydrate CRS.

B. It gives reaction (a) of bromides (2.3.1).





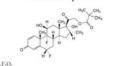


IDENTIFICATION

Example TLC

FLUMETASONE PIVALATE

Flumetasoni pivalas



M, 494.6

IDENTIFICATION

First identification: A, B.

Second identification: B, C, D.

A. Infrared absorption spectrophotometry (2.2.24). Comparison: flumetasone pivalate CRS.

If the spectra obtained in the solid state show differences, dissolve the substance to be examined and the reference substance separately in acctone R, evaporate to dryness on a water-bath and record new spectra using the residues.

B. Thin-layer chromatography (2.2.27).

Test solution. Dissolve 10 mg of the substance to be examined in acetone R and dilute to 10 mL with the same solvent.

solvent.

Reference solution (a). Dissolve 10 mg of flumetasone pivulate CRS in acetone R and dilute to 10 mL with the same solvent.

Reference solution (b). Dissolve 10 mg of desoxycortone acetate CRS in acetone R and dilute to 10 mL with the same solvent. Dilute 5 mL of this solution to 10 mL with reference solution (a).

reference solution (a).)
Plate: TLC silica gel F_{1st} plate R.
Mobile phase: add a mixture of 1.2 volumes of water R and
8 volumes of methanol R to a mixture of 15 volumes of
ether R and 77 volumes of methylene chloride R.

Application: 5 µL.

Development: over a path of 15 cm.

Drying: in air.

Detection A: examine in ultraviolet light at 254 nm. Results A: the principal spot in the chromatogram obtained with the test solution is similar in position and size to the principal spot in the chromatogram obtained with reference solution (a).

Detection B: spray with alcoholic solution of sulfuric acid R. Heat at 120 °C for 10 min or until the spots appear. Allow to cool. Examine in daylight and in ultraviolet light at 365 nm. cool. Examine in daylight and in ultraviolet light at 365 mm. Results B: the principal spot in the chromatogram obtained with the test solution is similar in position, colour in daylight, fluorescence in ultraviolet light at 365 mm and size to the principal spot in the chromatogram obtained with reference solution (a).

System suitability: reference solution (b):

the chromatogram shows 2 clearly separated spots.



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IDENTIFICATION

Example NMR

07/2019:0828 corrected 10.0

HEPARINS, LOW-MOLECULAR-MASS

Heparina massae molecularis minoris

IDENTIFICATION

A. Nuclear magnetic resonance spectrometry (2.2.33). Preparation: dissolve 0.200 g of the substance to be examined in a mixture of 0.2 mL of deuterium oxide R and

0.8 mL of water R. Comparison: dissolve 0.200 g of the appropriate specific low-molecular-mass heparin reference standard in a mixture of 0.2 mL of deuterium oxide R and 0.8 mL of

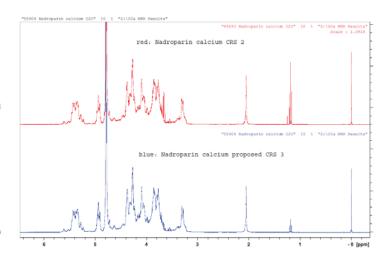
water R.

- Operating conditions:
 field strength: 75 MHz;
- temperature: 40 °C;
- cell diameter: 5 mm.

Processing:

- Fourier transformation;

 deuterated methanol reference signal set at 50.0 ppm.
 Results: the ¹³C NMR spectrum obtained is similar to that obtained with the appropriate specific low-molecular-mass heparin reference standard.







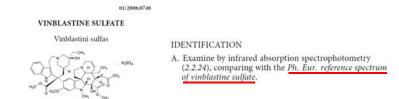
IDENTIFICATION

Technical guide for the elaboration of monographs (7th edition – 2015)

II.4.3. Infrared absorption spectrophotometry

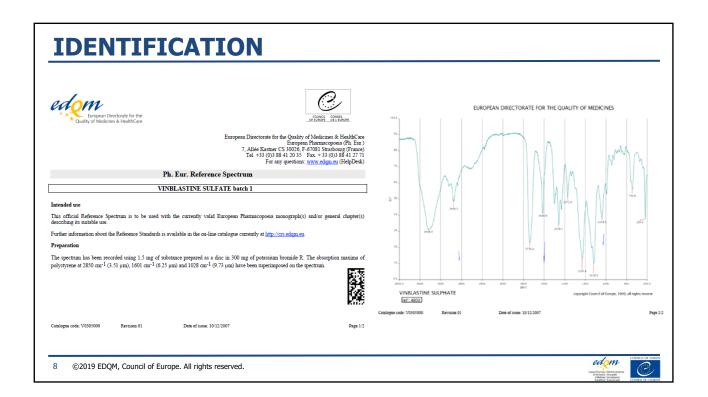
This method always necessitates the use of a **reference substance or a reference spectrum**. Reference substances are preferred to reference spectra. The latter are used where there are practical difficulties with providing a reference substance.

Note (not in technical quide): practical difficulties are e.q. toxicity, instability, risk for explosion.









ASSAY

Assay methods that generally require the use of an RS

- → liquid chromatography (LC) / gas chromatography (GC)
- (→ direct UV absorption spectrophotometry = to be phased out for assay: use specific absorbance)

RS strategy

- → whenever possible, substance compliant with corresponding Ph.Eur. monograph
- → higher amount of candidate material is required: extensive characterisation and increased amount per vial (sufficient for preparation of two solutions)
- → content is assigned based on compendial tests + complementary tests (if needed)
- → uncertainty of the assigned value shown to be negligible compared to content limits in the monograph

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ASSAY

RS strategy (continued)

- → RS may also be used for identification
- → other uses e.g. system suitability and peak identification are normally avoided
- → in justified cases, a lyophilised RS or a different salt or hydrate is used e.g. due to instability, hygroscopicity, ...





ASSAY

Example (assay RS = monograph substance)

04/2018:2887

RALTEGRAVIR POTASSIUM

Raltegravirum kalicum

C₂₀H₂₀FKN₆O₅ [871038-72-1]

M, 482.5

DEFINITION

Potassium 4-[[(4-fluorophenyl)methyl]carbamoyl]-1-methyl-2-[2-[(5-methyl-1,3,4-oxadiazol-2-yl)formamido]propan-2-yl]-6-oxo-1,6-dihydropyrimidin-5-olate.

Content: 98.0 per cent to 102.0 per cent (anhydrous substance).

ASSAY

Liquid chromatography (2.2.29) as described in the test for related substances with the following modification.

Injection: test solution and reference solution (a).

Calculate the percentage content of $\rm C_{20}H_{20}FKN_6O_5$ taking into account the assigned content of *raltegravir potassium CRS*.

INFORMATION LEAFLET Ph. Eur. Reference Standard Raltegravir potassium CRS batch 1

2.2 Analytical information related to intended use, when applicable

ne "as is" content is

: 99.1 % of C20H20FKN6O5

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ASSAY

• Example (assay RS ≠ monograph substance)

07/2014:2372

ESOMEPRAZOLE MAGNESIUM TRIHYDRATE

Esomeprazolum magnesicum trihydricum

 $C_{34}H_{36}MgN_6O_6S_2,3H_2O$

M, 767.2

DEFINITION

Magnesium bis[5-methoxy-2-[(S)-[(4-methoxy-3,5-dimethylpyridin-2-yl)methyl]sulfinyl]-1*H*-benzimidazol-1-idel trihvdrate.

Content: 98.0 per cent to 102.0 per cent (anhydrous substance).

IDENTIFICATION

A. Infrared absorption spectrophotometry (2.2.24).

Comparison: esomeprazole magnesium trihydrate CRS.

ASSAY

Liquid chromatography (2.2.29).

Calculate the percentage content of $\rm C_{34}H_{36}MgN_6O_6S_2$ taking into account the assigned content of omeprazole CRS.

 $1\ g$ of ome prazole is equivalent to 1.032 g of esome prazole magnesium.

INFORMATION LEAFLET Ph. Eur. Reference Standard
Omeprazole CRS batch 5

2.2 Analytical information related to intended use, when applicable

The "as is" content is

: 99.9 % of C17H19N3O3S (for 2787, 2372 and 2374)





Thank you for your attention



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